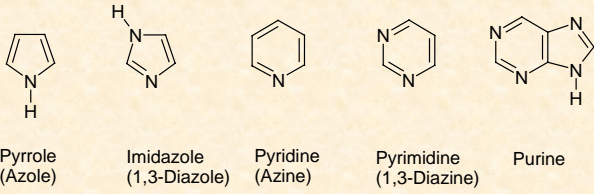


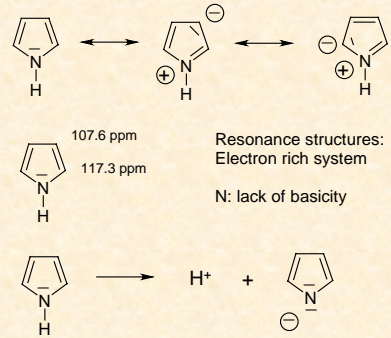
## Chemistry of nucleic acids

### 1. Basic properties of heterocycles

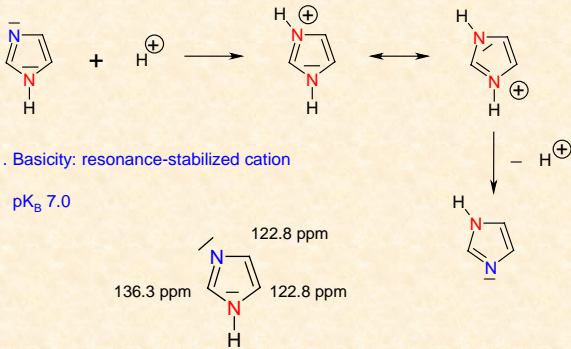


## Pyrrole

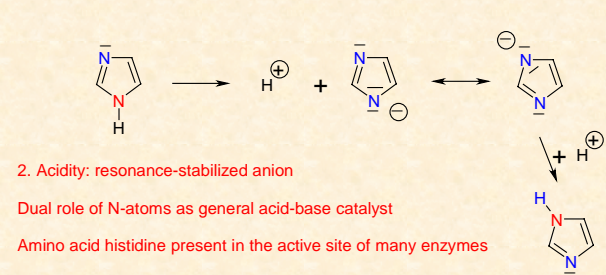
Pyrrole: Aromatic system (6  $\pi$ -Electrons)



## Imidazole

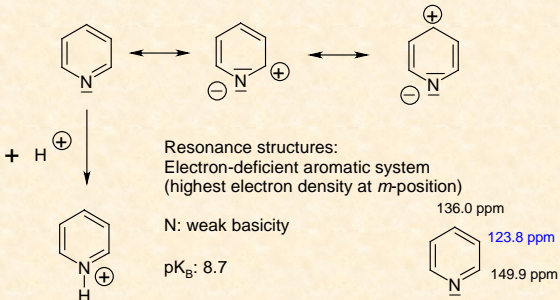


## Imidazole



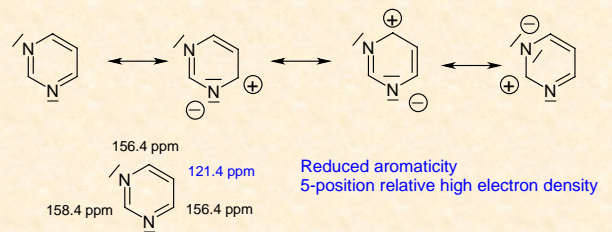
## Pyridine

Pyridine: 6  $\pi$ -electrons

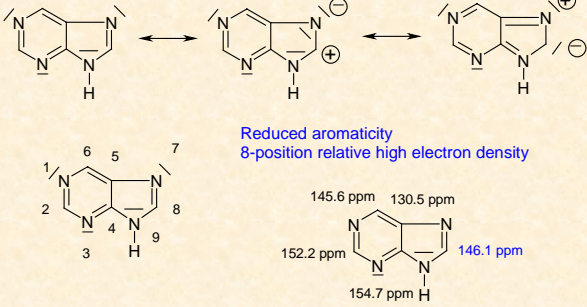


## Pyrimidine

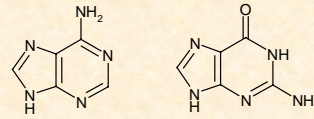
Pyrimidine: 6  $\pi$ -electrons



### Purine



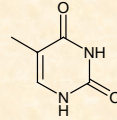
### N-Bases of DNA and RNA



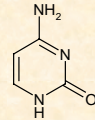
Adenine

Guanine

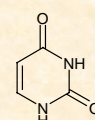
### Purin bases



Thymine



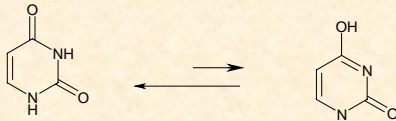
Cytosine



Uracil

### Pyrimidine bases

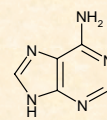
### Lactame-Lactime Tautomers



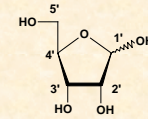
Uracil: Lactame form

Lactime form

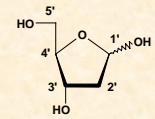
### Nucleoside: N-Glycoside of N-bases



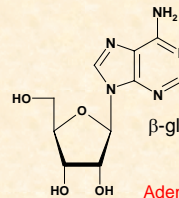
Adenine



Ribose  
(Furanose)



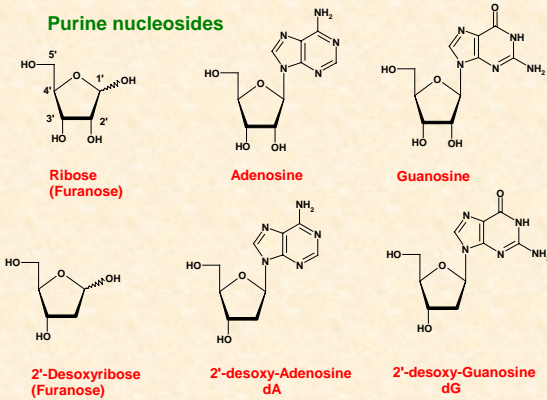
2'-Desoxyribose  
(Furanose)



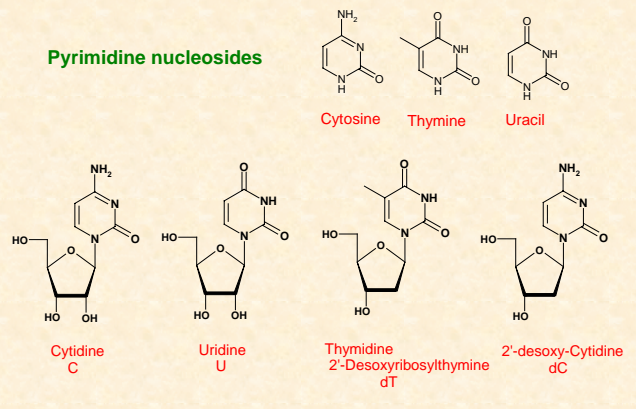
$\beta$ -glycosidic bond ( $\beta$ -ribofuranoside)

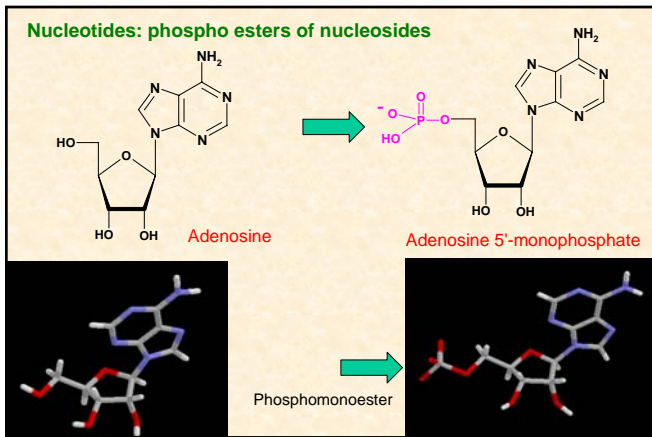
Adenosine

### Purine nucleosides



### Pyrimidine nucleosides





**2. General properties**

Solubility of pyrimidines and purines in water (1 g x g H<sub>2</sub>O)

Heterocycle	x g H <sub>2</sub> O	functional group
Pyrimidine	1	-
Uracil	280	2 OH
Thymine	250	2 OH, 1 Me
Purine.picrate	2	-
Adenine.picrate	1086	1 NH <sub>2</sub>
Guanine.picrate	26000	1 OH, NH <sub>2</sub>
2,8-Dihydroxyadenine	500000	1 OH, NH <sub>2</sub>
Uric acid	39480	3 OH
Nucleosides, Nucleotides	soluble in hot, insolub. in cold water	

**General properties**

Dissociation equilibria in nucleotides

Functional group	Dissociation reaction	pK
1. Phosphate-OH	$RO-PO_3H_2 \rightleftharpoons RO-PO_3H^- + H^+$	0.7-1.6
-NH <sub>2</sub> of adenine	$R-NH_3^+ \rightleftharpoons R-NH_2 + H^+$	3.5-4.5
2. Phosphate-OH	$RO-PO_3H^- \rightleftharpoons RO-PO_3^{2-} + H^+$	5.8-6.6
Heterocycl. protons (U, T, G)	$-NH-CO- \rightleftharpoons -N=C(O)- + H^+$	9.5
-OH of ribose	$R-OH \rightleftharpoons RO^- + H^+$	12.5

